

# Hamed Rasouli Sadabad

Dr Hamed Rasouli Sadabad is a postdoctoral research associate in InToxFIRE project in the FireSERT centre at Ulster University.

Hamed is a senior analytical chemist and materials scientist with education in Polymer Engineering (BSc, MSc) and Environmental Sciences (PhD).



# How Effective Are Gears at Blocking Chemical Exposure?

## What the Experiments Say

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# Why the Chemistry of Fire is Important?

Scope of  
the Study

Fire generates a wide range of chemicals, some of which are toxic, carcinogenic, irritant, etc.

The content and quantity of generated chemicals are not identical in fire events.

From the type of burning matter (e.g., wood, polymers, synthetic materials, batteries, etc. ) to ambient conditions (e.g., availability of O<sub>2</sub>, temperature, and relative humidity), and coincidence of various parameters and materials (e.g., presence of electricity), affect the concentration and kind of the generated toxicants.

Exposure to these chemicals is likely to impose loads of acute and long-term health problems to humans and other creatures as well as the whole environment.



# Why should we determine the transfer of chemicals through PPE?

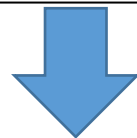
Firefighters and trainers are frequently exposed to these chemicals in different conditions, so they are more prone to the adverse effects of toxicants.



What is the concern? They wear PPEs that protect them to a good level!



What does “a good level” mean? 99%? 95%? Is it durable? Same in various conditions?



Some of those Chemicals such as benzo ( $\alpha$ ) pyrene, benzene, formaldehyde are categorized as Known Carcinogenic (IARC Group 1)

Scope of the Study





# What did we aim to investigate experimentally?

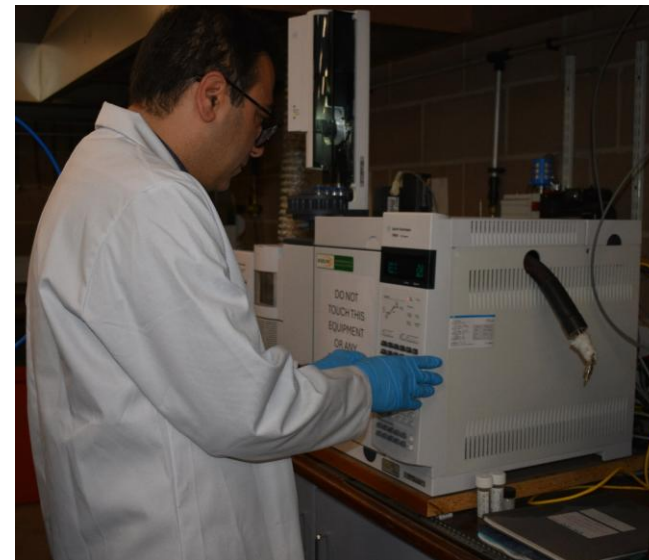
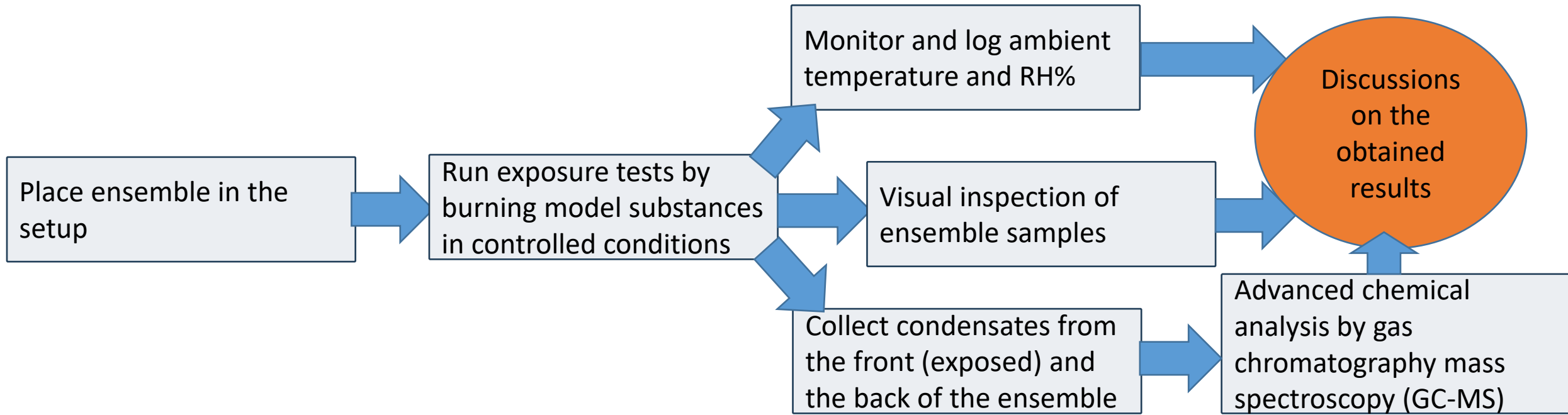
- To simulate the fire event conditions (*in the current presentation: small-scale laboratory setup*)
- To determine the nature and types of chemicals generated during fire events (*mainly focused on volatile organic compounds (VOCs) and polycyclic aromatic hydrocarbons (PAHs) in this presentation*)
- To evaluate the effectiveness of the studied ensembles in inhibiting different chemicals from passing through the ensemble towards fighter's body



InT<sup>re</sup>x<sup>fire</sup>FIRE



# Brief Protocol of Experiments



# Test Conditions and Initial Results

**Ensemble:** New,  
*With PFAS layer on its membrane (PFAS)*

*No PFAS layer on its membrane, water repellant on final layer*

**Fuel (10 g solid + 10 mL Hexane)**  
Wood: 40%, PU foam: 20%; PET: 20%; PS: 10%, PVA: 10%

**Air flow rate:** 400 L/min

**Suction pressure:** -25 mbar, immediately drops to 0 when connected, since air is soaked.

**Test time:** 3 \*15 min

**Residue:**  $39.5 \pm 2.5$  %

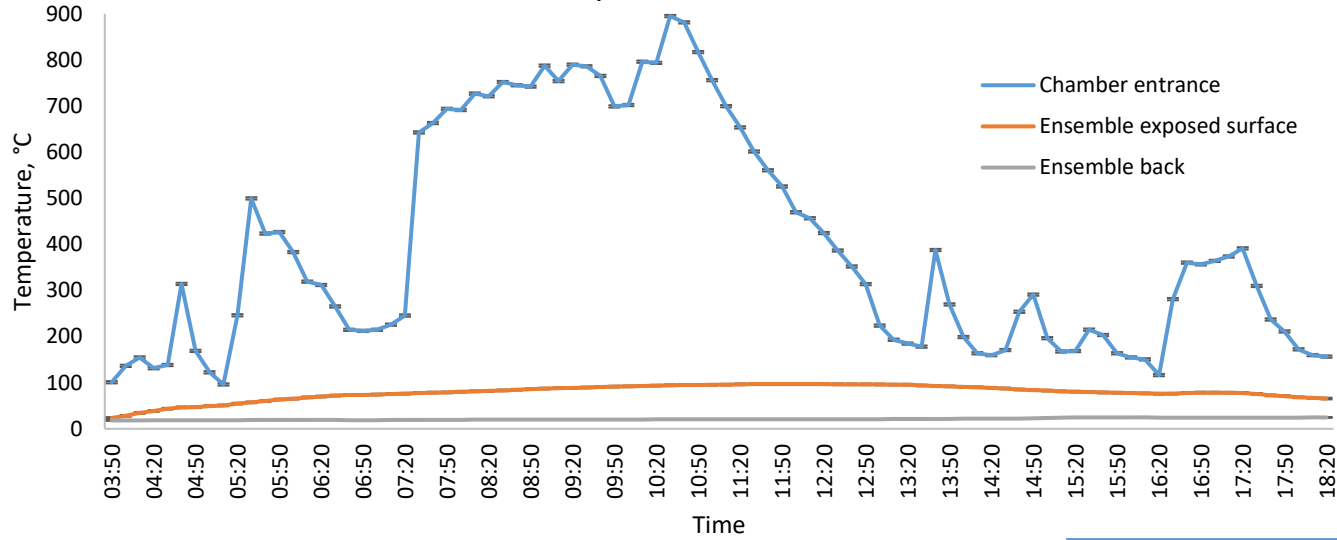
Fire Exposure experiments

Monitored item		PFAS layer on membrane	No PFAS layer on membrane
Released steam pH	Front (exposed)	2-3	2-3
	Back	Similar to reference	Similar to reference
Condensate pH	Front (exposed)	$2.5 \pm 0.1$	$2.5 \pm 0.1$
	Back	Similar to reference ( $6.0 \pm 0.1$ )	$5.0 \pm 0.1$
Relative humidity change	Back	Similar to reference ( $55 \pm 5\%$ )	From 55 (reference) to $65 \pm 5\%$



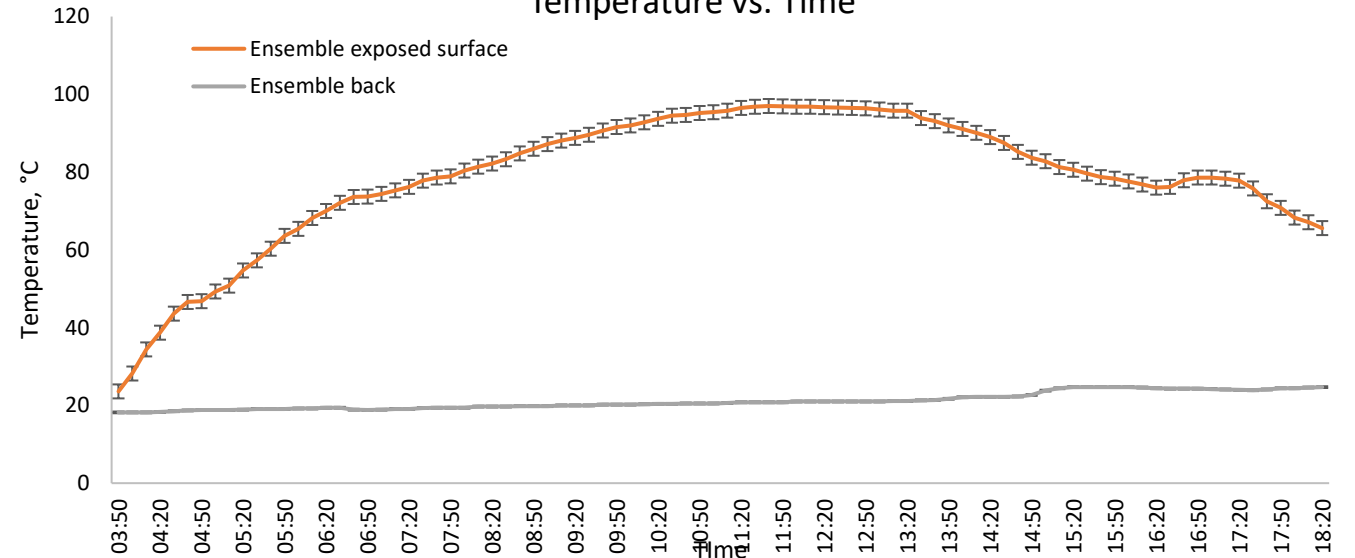
# Test Conditions and Initial Results

Temperature vs. Time



**Temperature  
change regime  
during the  
experiment**

Temperature vs. Time





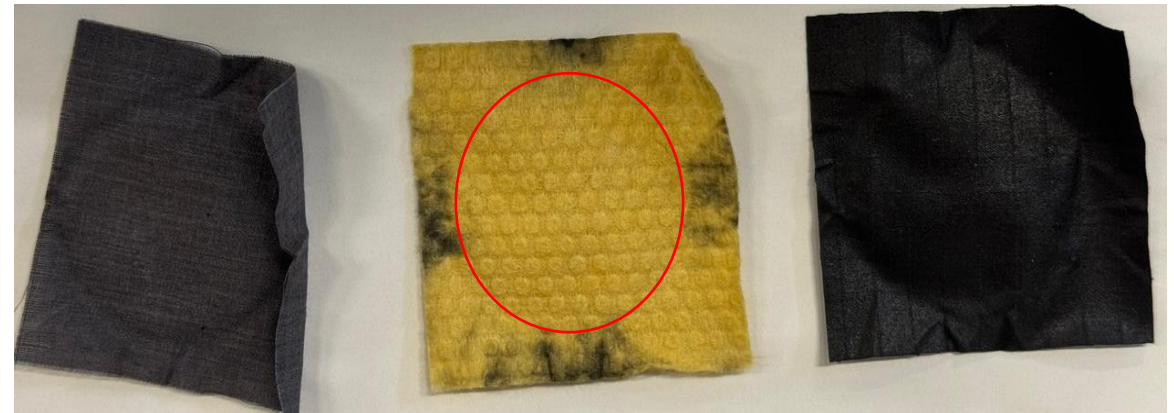
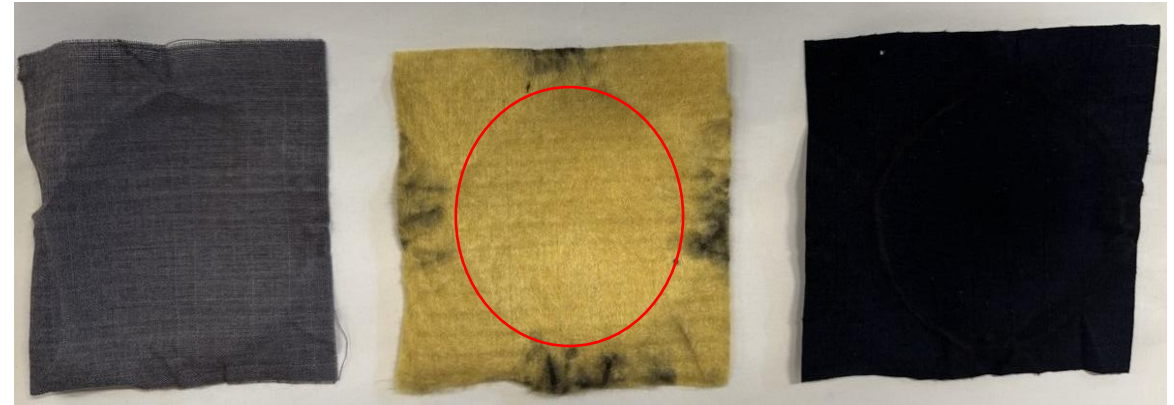
# Test Conditions and Initial Results

Visual inspection of the ensembles before and after exposure experiments

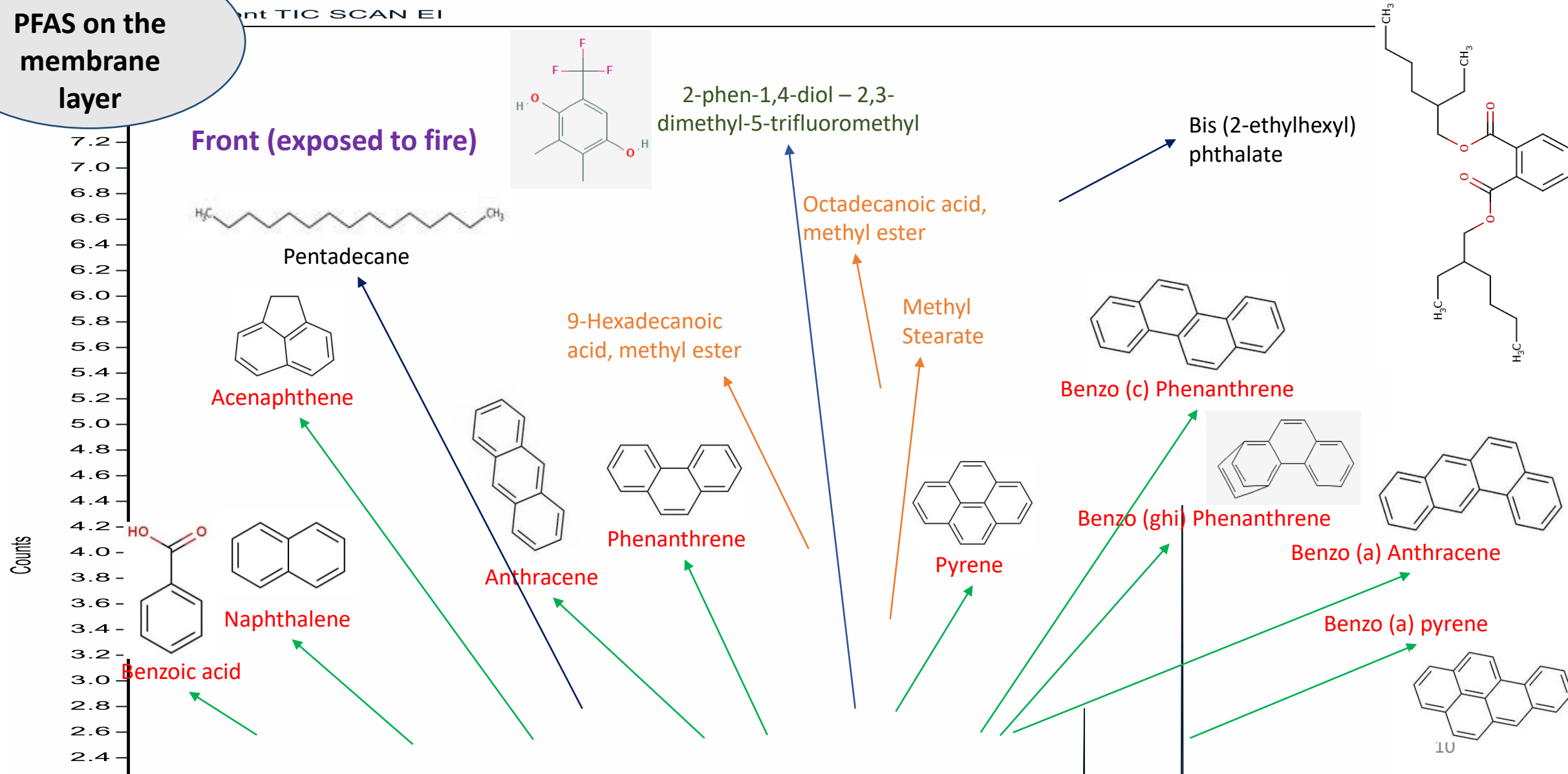
**Ensemble: with PFAS**



**Ensemble: without PFAS**



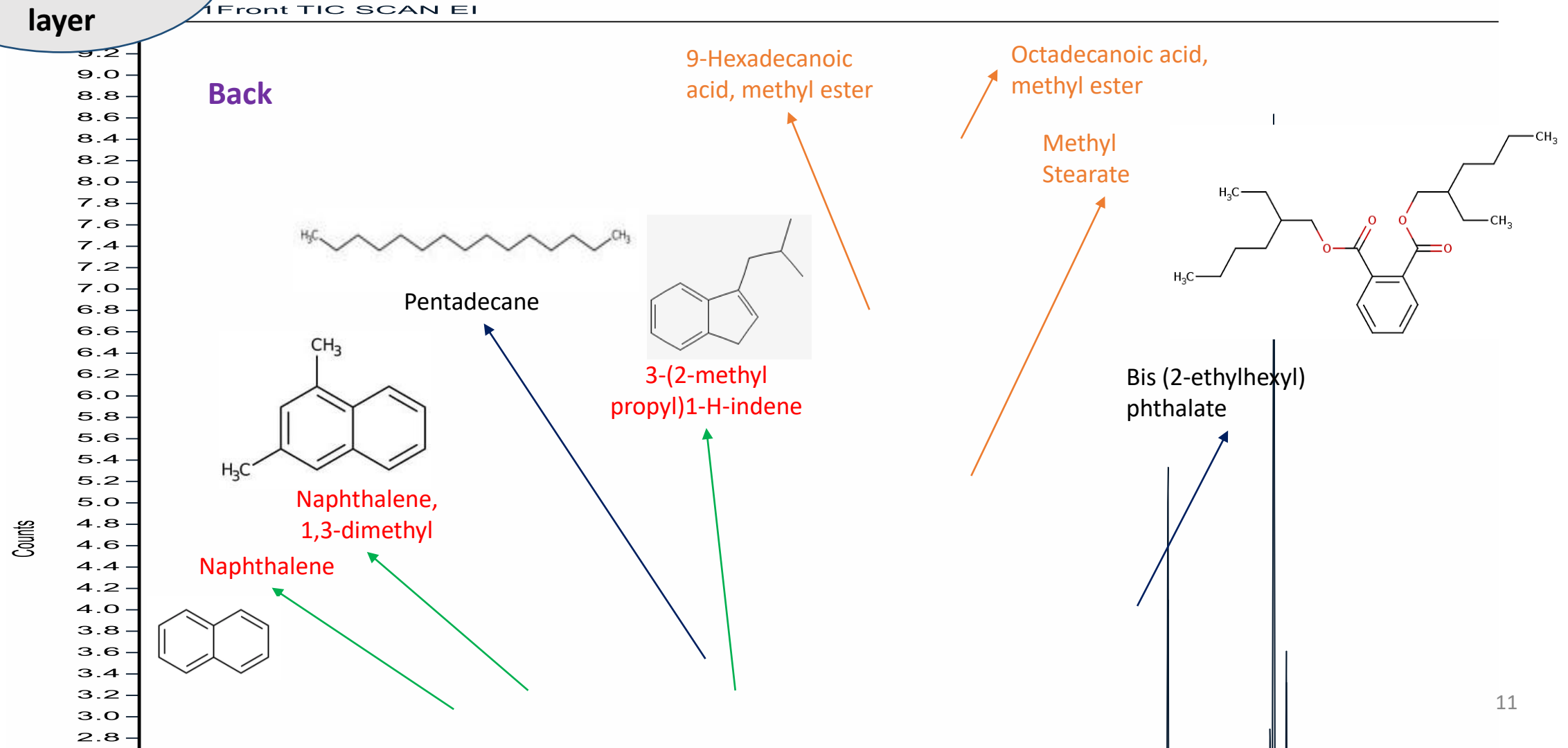
## Ensemble with PFAS on the membrane layer



# Advanced Analysis: Toxic Chemicals Detection

Ensemble with  
PFAS on the  
membrane  
layer

Chemical analysis



# Advanced Analysis: Toxic Chemicals Detection

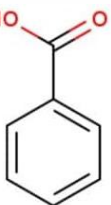
Ensemble no  
PFAS on the  
membrane  
layer

Front TIC SCAN EI

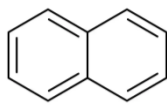
Front (exposed)

Counts

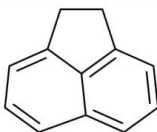
1.250  
1.225  
1.200  
1.175  
1.150  
1.125  
1.100  
1.075  
1.050  
1.025  
1.000  
0.975  
0.950  
0.925  
0.900  
0.875  
0.850  
0.825  
0.800  
0.775  
0.750  
0.725  
0.700  
0.675  
0.650  
0.625  
0.600  
0.575  
0.550  
0.525  
0.500  
0.475  
0.450  
0.425  
0.400



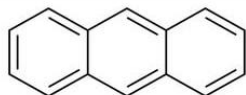
Benzoic acid



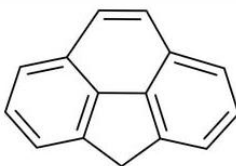
Naphthalene



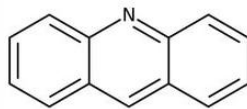
Acetanaphthene



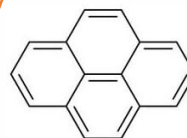
Anthracene



Cyclo-penta [def]  
phenanthrene



Acridine



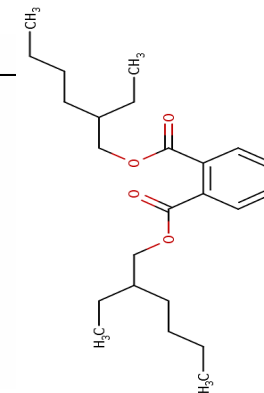
Pyrene

9-Hexadecanoic  
acid, methyl ester

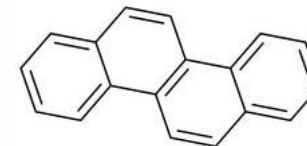
Octadecanoic acid,  
methyl ester

Methyl  
Stearate

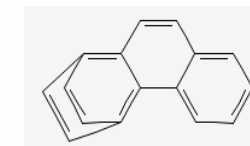
Bis (2-ethylhexyl)  
phthalate



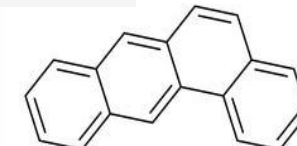
Benzo (c)  
fluorantine



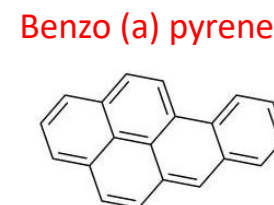
Benzo (c) Phenanthrene



Benzo (ghi) Phenanthrene



Benzo (a) Anthracene

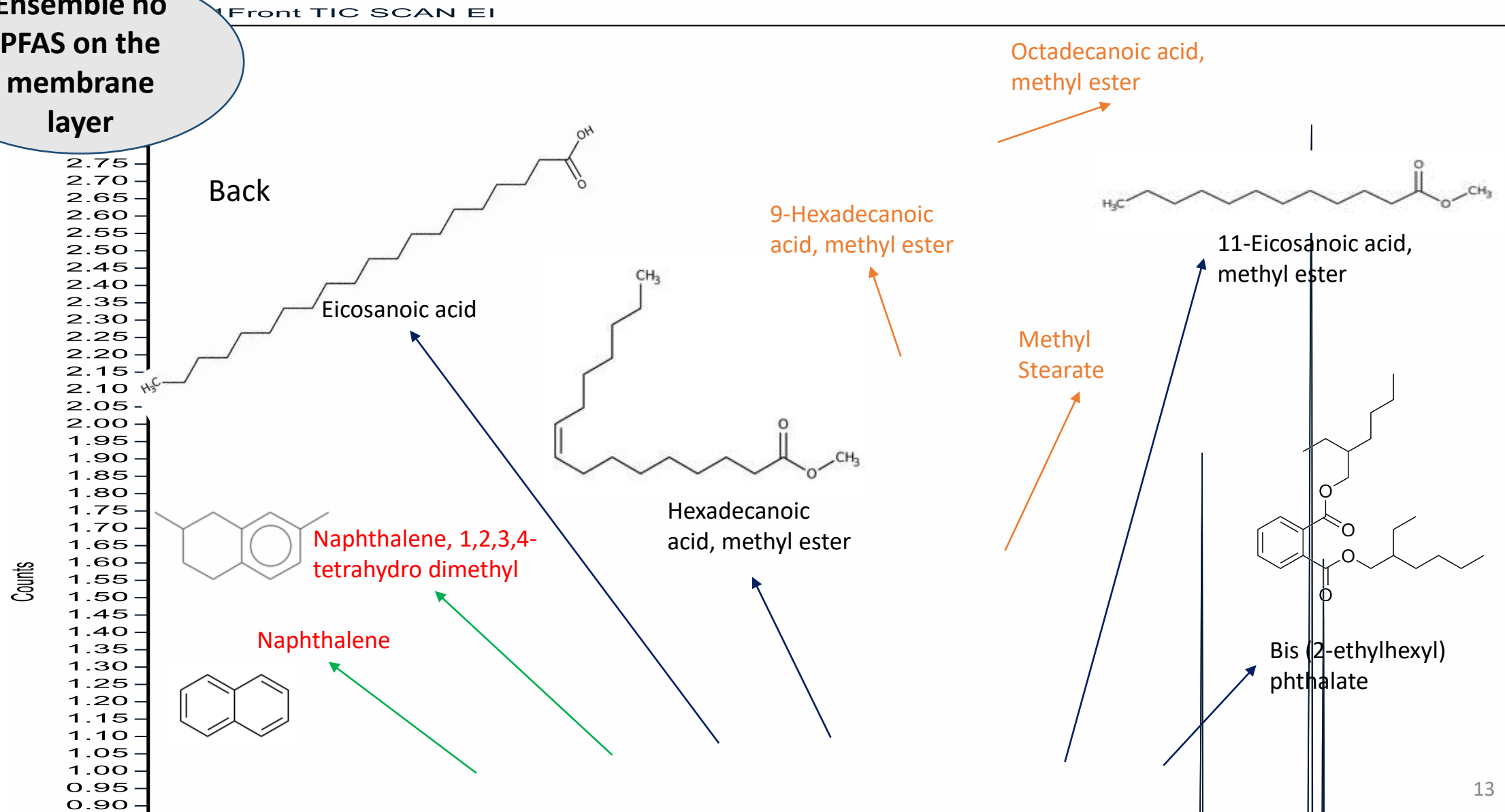


Benzo (a) pyrene



# Advanced Analysis: Toxic Chemicals Detection

Ensemble no  
PFAS on the  
membrane  
layer



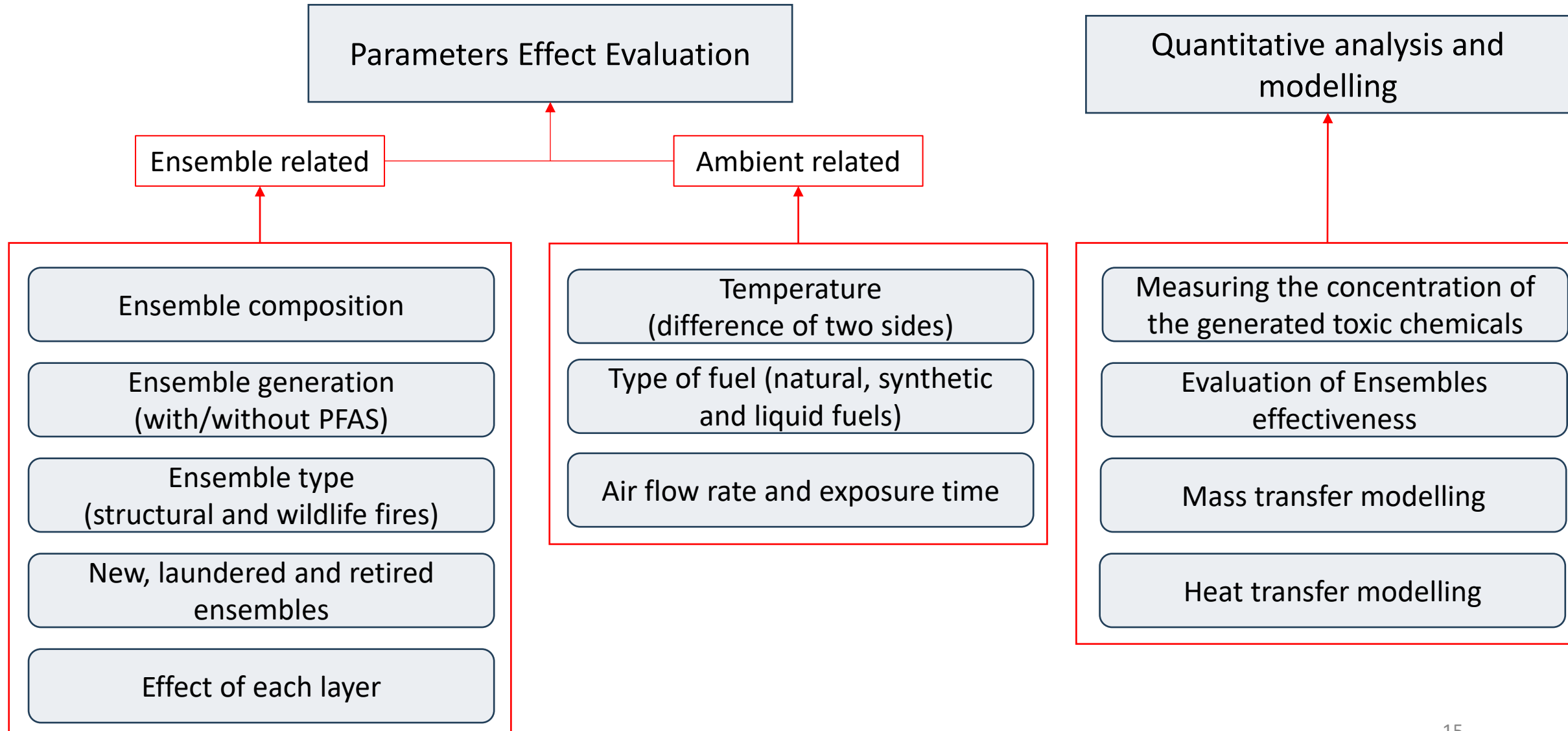
# Take Home Message from Experimental Results



## Conclusions

- Both ensembles are effective against diffusion of large PAHs.
- Naphthalene and its derivatives are able to diffuse to both ensemble, regardless of presence or absence of PFAS on the membrane layer.
- Linear chemicals are able to pass through the ensembles, even if they have a super-large molecule (eicosanoid acid, hexadecenoic acid methyl ester, etc.)
- Ensemble with PFAS on its membrane showed better barrier effect in stopping linear compounds to pass through it.
- Ensemble with PFAS on its membrane is likely to release fluoride-containing compound(s) to its exposed face to fire (Yet needs to be confirmed by more experiments).

# Next in Experimental Studies





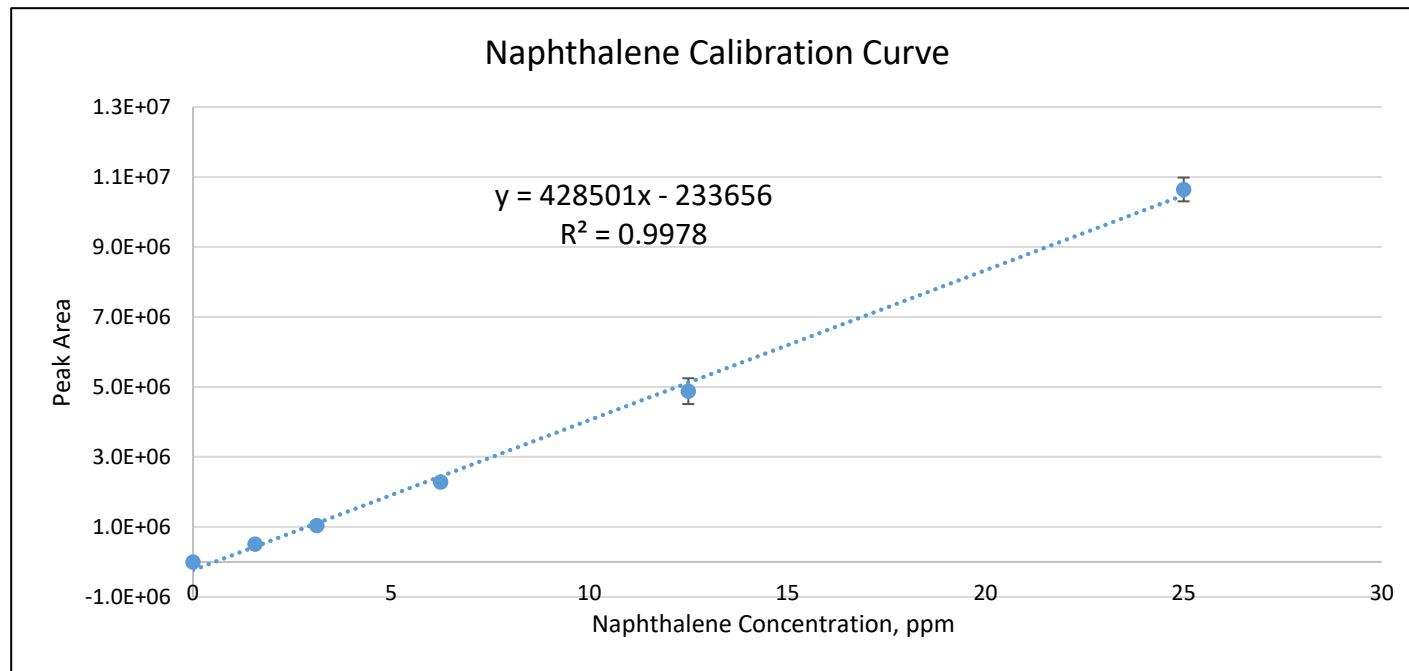
Thanks for your  
attention



# WP2: Small-Scale Experimental Studies

Quantitative analysis

Naphthalene



Item	Value (ppm)
Limit of Quantification (LOQ)	1.5625
Limit of Detection (LOD)	0.3906

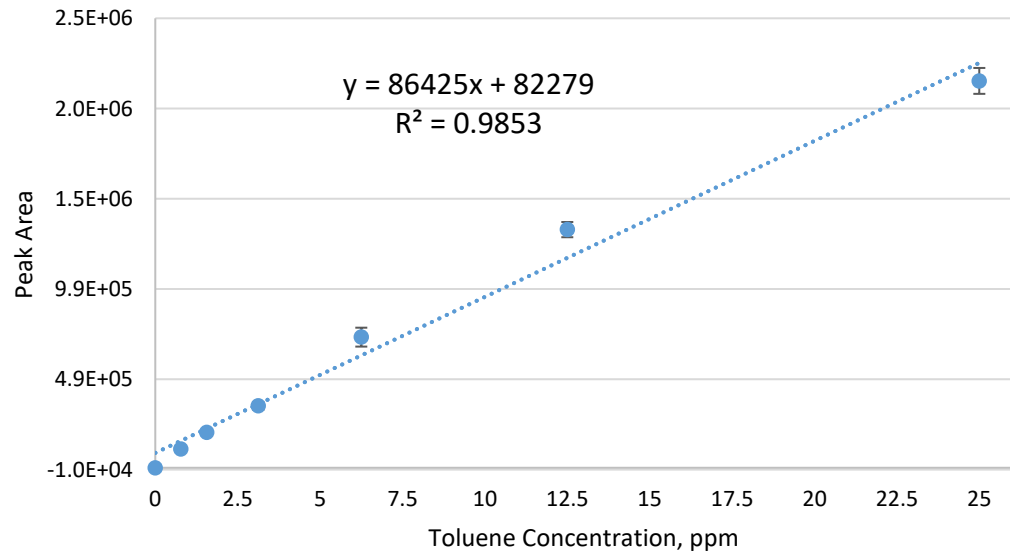
# WP2: Small-Scale Experimental Studies

Quantitative analysis

Toluene

Item	Value (ppm)
Limit of Quantification (LOQ)	1.5625
Limit of Detection (LOD)	0.1950

Toluene Calibration Curve



Toluene Calibration Curve

